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In your lab notebook decide what data you will need to collect in order to answer the research question. Develop your procedures and decide how you will collect your data. Perform the virtual experiment and analyze your results. Develop a scientific argument (claim, evidence, reasoning) that answers the research question.

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#### Mr Palermo Spectroscopy Lab Answers

Conclusion: (answer on your lab sheet) 1. What evidence is there that electrons move around in definite pathways around the nucleus? 2. How is emission spectra produced? 3. How might emission spectra be used in studying stars? 4. Draw a Bohr diagram for Hydrogen and Neon. (If you need help on drawing Bohr diagrams click on this link below)

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Lab Requirements: As a group, make a prediction for the research question. As a group, develop a model/explanatory hypothesis that you will test through experimentation to develop your claim. Remember to get feedback on your model/explanatory hypothesis from other groups and Mr. Palermo.

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#### Emission Spectroscopy Lab Answer Key

Students work to match up emission spectra from the question cards with the known element emission spectra and record their answers on a sheet of paper. As students are working, I am accessible and moving from station to station to observe the student interactions and answer any questions.

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#### Spectroscopy Part I - Langara College

Click the link to check your answers: Answers Objective: The aim of this lab is to practice recording measurements with the correct precision. Procedure: Several stations have been set up in this virtual lab. For each station, click on the measuring tool link, read the directions, and read the measurements according to the directions.

#### Virtual Lab Precision and Significant Figures - Mr ...

In 1905, Einstein found the answer: Light is both! In some situations it behaves like waves, while in others it behaves like particles. This may seem odd. How can light act like both a wave and a particle at the same time? ... Emission Spectroscopy of Elements - Simulation: ... FLAME TEST AND ATOMIC SPECTRA LAB ...

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1. Read the introduction and answer the following pre-lab questions. a. Define spectroscopy. b. Which color light has the shortest wavelength? c. Which color light has the longest wavelength? d. Compare and contrast the 3 types of spectra. 2. Follow the directions for the Flame test lab and record your data in the appropriate table below. a.

#### Title of Lab - Easy Peasy All-in-One High School

Analysis of this light can be used to identify elements and molecules. In today's lab, we will observe the visible range of light emission. Objectives 1. Perform a flame test to identify the characteristic color of metal ions. 2. Identify the metal ion in an unknown solution. 3. Calculate the energy of emitted photons. Materials: (per lab group)

#### Flame Test Lab Activity Key - University of South Florida

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Lab Requirements: As a group, make a prediction for the research question. As a group, develop a model/explanatory hypothesis that you will test through experimentation to develop your claim. Remember to get feedback on your model/explanatory hypothesis from other groups and Mr. Palermo.

Other Results for Emission Spectroscopy Lab Answer Key: Ninth grade Lesson Element Emission Spectra | BetterLesson. Students work to match up emission spectra from the question cards with the known element emission spectra and record their answers on a sheet of paper.

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