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$[C]^x[D]^y [A]^n[B]^m$ 556 CHAPTER 18
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The ion product exceeds the K_{sp} .

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The ion product exceeds the K_{sp} .

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CHAPTER 18 Chemical Equilibrium

1. In a bottle of unopened cola, the CO₂ gas dissolved in the liquid is in equilibrium with the CO₂ gas above the liquid. The dissolved gas reacts with water molecules in the cola to form carbonic acid, which also dissociates into carbon dioxide and water. Which chemical equation(s) best describe this equilibrium system?

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